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The Novel Multidentate Ligands ${}^t\text{Bu}_2\text{P}(2\text{-SePy})$ and ${}^t\text{BuP}(2\text{-SePy})_2$; Py = Pyridyl. Synthesis, Reactivity and Crystal Structures of Mo-, Cu- and Ag-Complexes

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2-Pyridylseleno-phosphines can easily be synthesized via reaction of appropriate halophosphines with $\text{K}(2\text{-SePy})$ or $[\text{Et}_4\text{N}][2\text{-SePy}]$. The phosphines are extremely sensitive to oxygen and can be trapped as Cu-, Ag- or Mo-complexes via reaction with CuBr, AgBr and $[(\eta^6\text{-C}_7\text{H}_8)\text{Mo}(\text{CO})_3]$. The crystal structures of ${}^t\text{BuP}(\text{O})(2\text{-SePy})_2$ **7**, $[\{{}^t\text{Bu}_2\text{P}(2\text{-SePy})\}\text{-P}, N\text{-MBr}]_2$, M = Cu **8**, Ag **9** and $[\{{}^t\text{BuP}(2\text{-SePy})_2\text{-N}, N', P\text{-Mo}(\text{CO})_3\}]$ **10** have been determined.

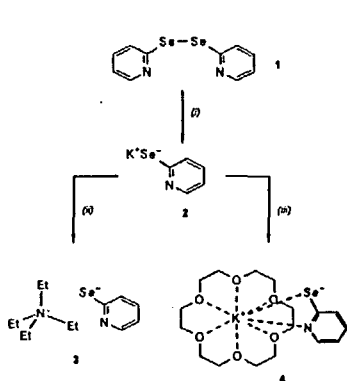
Keywords: Pyridylselenolate; phosphine; metal complex

The current interest in the chemistry of heteroatom-substituted phosphines arises from their potential use in catalytical processes. One possibility of controlling the reaction pathways and the selectivity of phosphine-containing catalysts is the modification of the surrounding phosphine ligands by varying the donor-atom sets, which are usually C, N and P[1,2].

Here, we describe the synthesis and reactivity of novel 2-pyridylseleno-phosphine ligands that contain N-, P-, Se-donor sets. Recently, a few complexes containing the SePy^- anion appeared in the literature, mainly in combination with Cd, Hg and Pd, as single-source precursors for metal selenides[3]. Starting from 2,2'-bis(pyridyl)diselenide[4] **1**, the SePy^- anion can easily be obtained as depicted in Scheme 1. Reactions with $\text{K}(2\text{-SePy})$ **2** are usually performed *in situ*. $[\text{Et}_4\text{N}][2\text{-SePy}]$ **3** is a useful derivative, because it can be stored under N_2 for a prolonged time without decomposition[5].

$^t\text{Bu}_2\text{P}(2\text{-SePy})$ **5** and $^t\text{BuP}(2\text{-SePy})_2$ **6** can easily be prepared from halophosphines via reactions with **2** or **3** (Scheme 2)[6] in nearly quantitative yield (for **6** by NMR means).

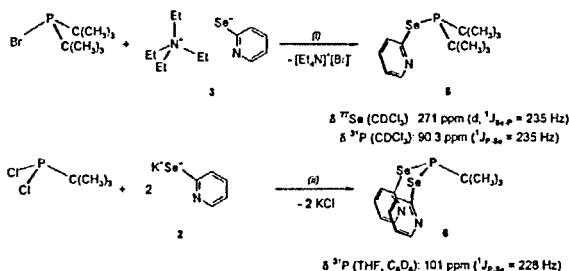
5 and **6** are extremely air-sensitive oily liquids (**5** solidifies at 25°C). During the work-up procedure, **6** was fully oxidized to the corresponding phosphine oxide $^t\text{BuP}(\text{O})(2\text{-SePy})_2$ **7** (Fig. 2), which was characterized by spectroscopic and crystallographic



SCHEME 1: Reaction conditions: (i) KSCN , Bu_3H , THF, 25°C ; (ii) $[\text{Et}_4\text{N}]\text{Br}$, EtOH, 25°C ; (iii) 18-crown-6, EtOH, 25°C .

means [$\delta^{77}\text{Se}$ (CDCl_3): 438 ppm ($^1\text{J}_{\text{Se-P}}$ 418 Hz); $\delta^{31}\text{P}$ (D_6 -acetone): 72.1 ppm ($^1\text{J}_{\text{P-Se}}$ 413 Hz)] [6]. **7** represents the first crystal structure of a bis(pyridylseleno)phosphine oxide. The observed P-Se bond lengths are attributable to P-Se single bonds

and comparable to the central P-Se bonds in $[^1\text{BuP}(\text{Se})(\mu\text{-Se})_2]$ (2.269(2) Å)[7]. **5** and **6** were trapped as Cu-, Ag- and Mo-complexes via reactions with CuBr, AgBr or $[\eta^6\text{-C}_7\text{H}_8]\text{Mo}(\text{CO})_3$; tht = tetrahydrothiophene, C_7H_8 = cyclooctatriene) in excellent yield[6]. The reactions were performed under exclusion of light.



SCHEME 2: Synthesis of pyridylseleno-phosphines, reaction conditions: (i) CH_3CN , rt, 1h; (ii) THF, rt, 3h.

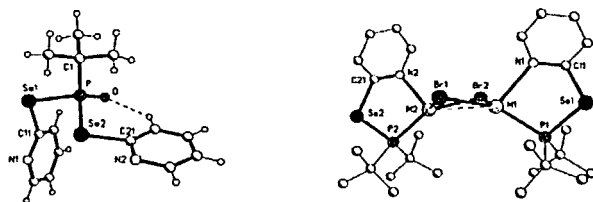


FIGURE 1: The crystal structures of **7** (left, $C2/c$, $Z = 8$) and the crystal structures of **8** and **9** (right, $P-1$, $Z = 2$) displaying the five-membered MPSeCN-rings. Selected bond lengths [Å] and angles [°]: **7** P-Se1 2.2468(11), P-Se2 2.2562(12), P-O 1.470(3), Se1-C11 1.933(4), Se2-C21 1.951(4); C11-Se1-P 95.28(12), C21-Se2-P 104.92(12). **8** [$M = \text{Cu}$] Cu-P1 2.2015(17), P1-Se1 2.2641(17), Se1-C11 1.918(6), C11-N1 1.344(6), Cu-N1 2.109(4), Cu-Br1 2.4596(10), Cu-Br2 2.4977(10), Cu...Cu 3.1430(11), Br...Br 3.7766(10); N1-Cu1-P1 95.11(14), Cu-P1-Se1 101.07(7), P1-Se1-C11 99.31(17). **9** [$M = \text{Ag}$] Ag-P1 2.4116(11), P1-Se1 2.2545(12), Se1-C11 1.945(4), C11-N1 1.325(5), Ag-N1 2.456(3), Ag-Br1 2.6312(6), Ag-Br2 2.7017(6), Ag...Ag 3.1333(6), Br...Br 4.2153(8); N1-Ag-P1 85.14(8), Ag-P1-Se1 105.33(5), P1-Se1-C11 100.98(12).

The molybdenum complex is shown in Fig. 2. Mo is coordinated octahedrally with a fac-pyridylseleno-phosphine ligand.

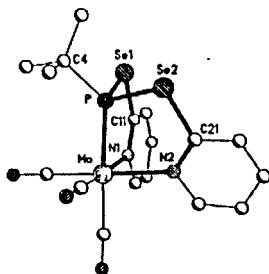


FIGURE 2: The molecular structure of 10. Selected bond lengths [Å] and angles [°]: Mo-P 2.3924(10), P-Se1 2.2659(10), P-Se2 2.2712(10), Mo-N 2.302(3) av., Se-C 1.920(4) av.; Se1-P-Mo 108.01(4), Se2-P-Mo 106.21(4), C4-P-Mo 132.32(12), C4-P-Se1 101.31(12), C4-P-Se2 101.26(12), P-Se1-C11 97.79(11), P-Se2-C21 96.28(11).

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References

- [1] J. P. Collman, L. S. Hegedus, J. R. Norton, R. G. Finke, *Principles and Applications of Organotransition Metal Chemistry*, University Science Book, CA, (1987).
- [2] a. A. Togni, L. M. Venanzi, *Angew. Chem., Int. Ed. Engl.*, **33**, 497, (1994). – b. H. A. Mayer, W. C. Kaska, *Chem. Rev.*, **94**, 1239, (1994).
- [3] S. Narayan, V. K. Jain, B. Varghese, *J. Chem. Soc., Dalton Trans.* 2359, (1998) and references cited therein.
- [4] C. O. Kienitz, C. Thöne, P.G. Jones, *Inorg. Chem.*, **35**, 3990 (1996).
- [5] J. Laube, S. Jäger, C. Thöne, *Chem. Commun.*, (2000), submitted.
- [6] J. Laube, *PhD thesis*, Braunschweig (2000).
- [7] J. T. Shore, W. T. Pennington, M. C. Noble, A. W. Cordes, *Phosphorus, Sulfur and Silicon*, **39**, 153, (1998).